CHM 1143

Homework Set 2

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1)	When excited, sodium atoms emit an intense yellow color with a wavelength of 589 nm. What is the energy of these 589 nm photons?
2)	What is the molarity of a solution prepared by dissolving 14.0 g of methanol CH_3OH) in enough water to make 450 mL of solution?
3)	How many moles of glucose ($C_6H_{12}O_6$) are contained in 575 mL of a 0.25 M aqueous glucose solution?
4)	What is the concentration of glucose in a solution prepared by adding 47 mL of a 0.33 M glucose to enough water to make 1.00 L of solution?
5)	How many mL of 16 M HNO $_{\rm 3}$ must be used to prepare 500. mL of a 0.25 M solution?